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elements. NaHCO₃ g Na₂ CO₃ + H₂
O + CO₂ 7. The names are found For

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A Voyage through Equations ANSWER

KEY Section 1: Identify the type of
reaction 1) ... Practicing equation

balancing 1) $2 \text{C}_6\text{H}_6 + 15 \text{O}_2 \rightarrow 6 \text{H}_2\text{O} + 12 \text{CO}_2$ 2) $4 \text{NaI} + 1 \text{Pb}(\text{SO}_4)_2 \rightarrow 1$

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PbI₄ + 2 Na₂SO₄ ... Predicting the products of chemical reactions 1) 2
Ag + 1 CuSO₄ 1 Ag₂SO₄ + 1 Cu Type:
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Equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not the subscript.

49 Balancing Chemical Equations
Worksheets [with Answers]

Page 1. Balancing Equations

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Worksheet – Answers. Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number "1".

$1 \text{ Na}_3\text{PO}_4 + 3 \text{ KOH} \rightarrow 3 \text{ NaOH} + 1 \text{ K}_3\text{PO}_4$

$1 \text{ MgF}_2 + 1 \text{ Li}_2\text{CO}_3 + 1 \text{ MgCO}_3 + 2 \text{ LiF}$

$1 \text{ P}_4 + 3 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_3$

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Chemical reactions have the same number of atoms before the reaction as after the reaction. Balancing chemical equations is a basic skill in

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